

# Measurement and modelling concentrations and dry deposition of ammonia: methods and challenges

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*and many other colleagues*

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# OVERVIEW

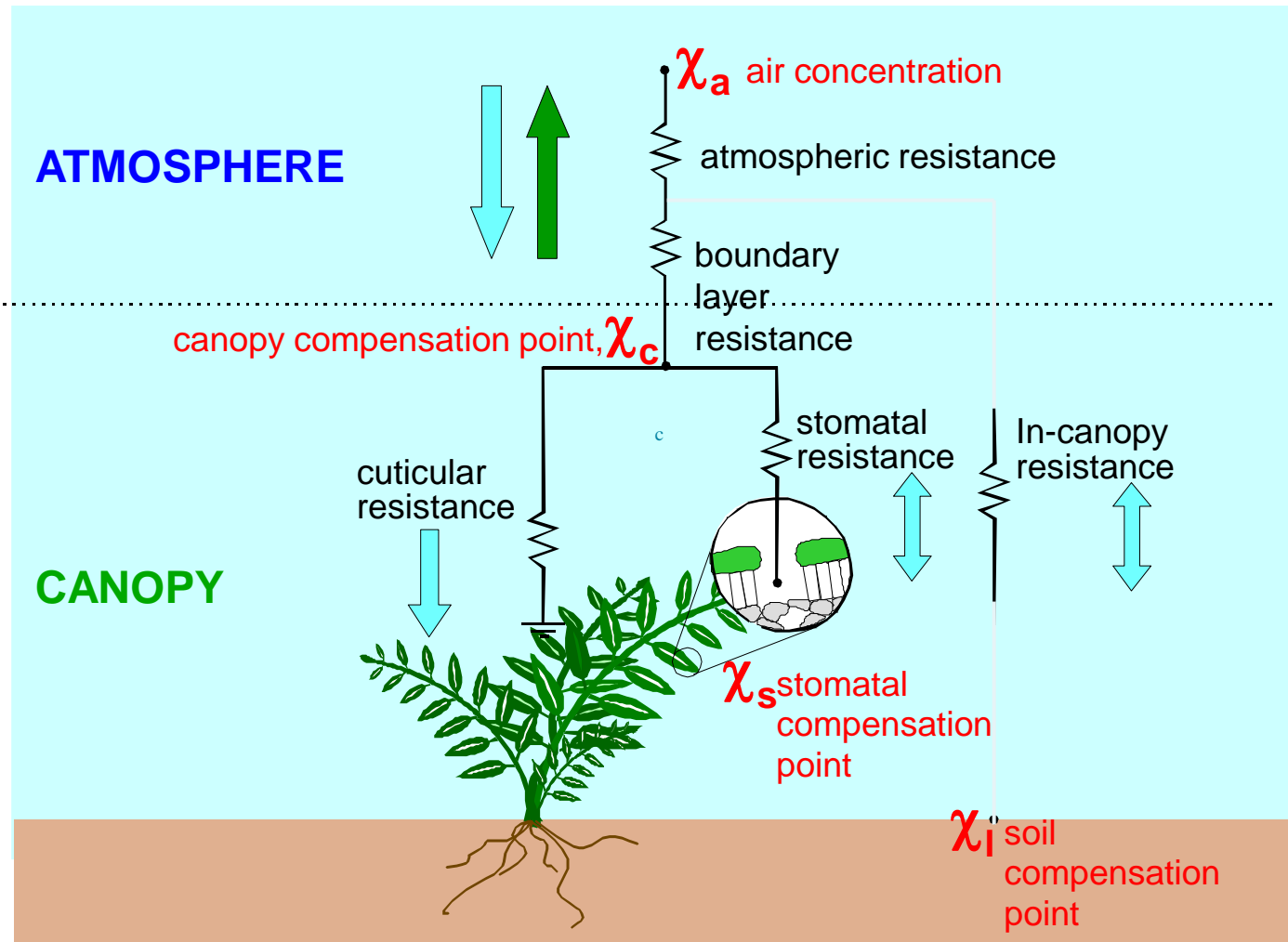
- **Modelling of dry deposition of  $\text{NH}_3$**
- **Measurement of dry deposition of  $\text{NH}_3$**
- **Current  $\text{NH}_3$  monitoring in the UK**
- **Challenges in quantifying  $\text{NH}_3$  deposition**
- **Ammonia as a future fuel: some considerations**

# Canopy Compensation Point Model

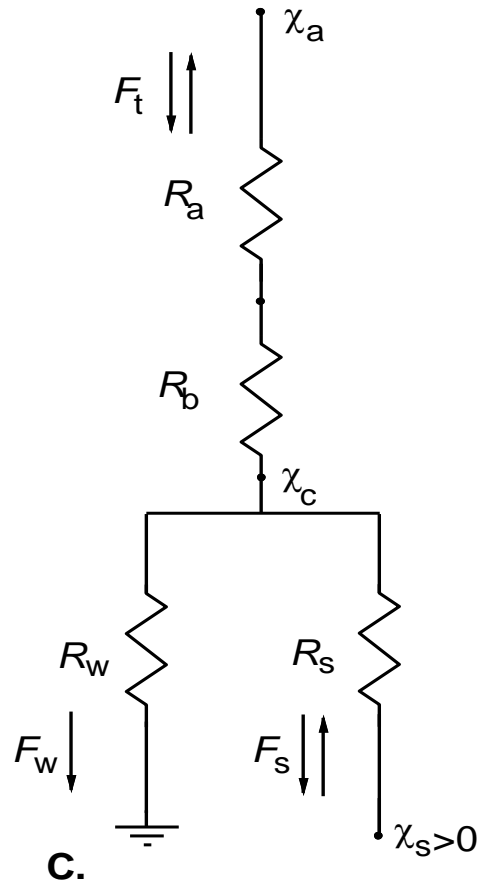
$$\chi_s = f(T, \Gamma)$$

$$\Gamma = \frac{[\text{NH}_4^+]_{\text{apo}}}{[\text{H}^+]_{\text{apo}}}$$

Nemitz, Milford & Sutton  
(Quart. J. Royal Meteor.  
Soc. 2000)



# Key Parameters of the Canopy Compensation Point Model



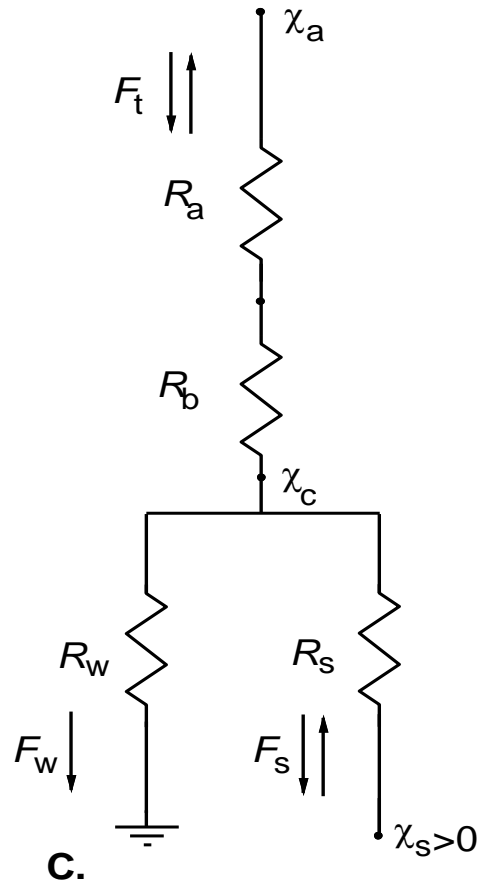
- Atmospheric resistances: easy! from micromet parameters ( $u_*$ ,  $U$ ,  $L$ ,  $z_0$ )
- Stomatal resistance  $R_s$ : function of radiation (St, PAR), leaf water potential, RH/VPD, temperature
- Cuticular resistance  $R_w$ : function of leaf wetness (RH, VPD, leaf wetness measurement), leaf water composition (pH), temperature?, LAI?
- Dimensionless stomatal emission potential  $\Gamma_s = [\text{NH}_4^+]/[\text{H}^+]$ ; function of N status, plant type

$$\chi_s = \text{NH}_{3(g)} = \frac{161500}{T} \exp(-10,378 T^{-1}) \frac{[\text{NH}_4^+]}{[\text{H}^+]} = \frac{161500}{T} \exp(-10,378 T^{-1}) \Gamma_s$$

$$F_t = -\frac{\chi_a - \chi_c}{R_a(z-d) + R_b}$$

$$\chi_c = \frac{\chi_a (R_a(z-d) + R_b)^{-1} + \chi_s R_s^{-1}}{(R_a(z-d) + R_b)^{-1} + R_s^{-1} + R_w^{-1}}$$

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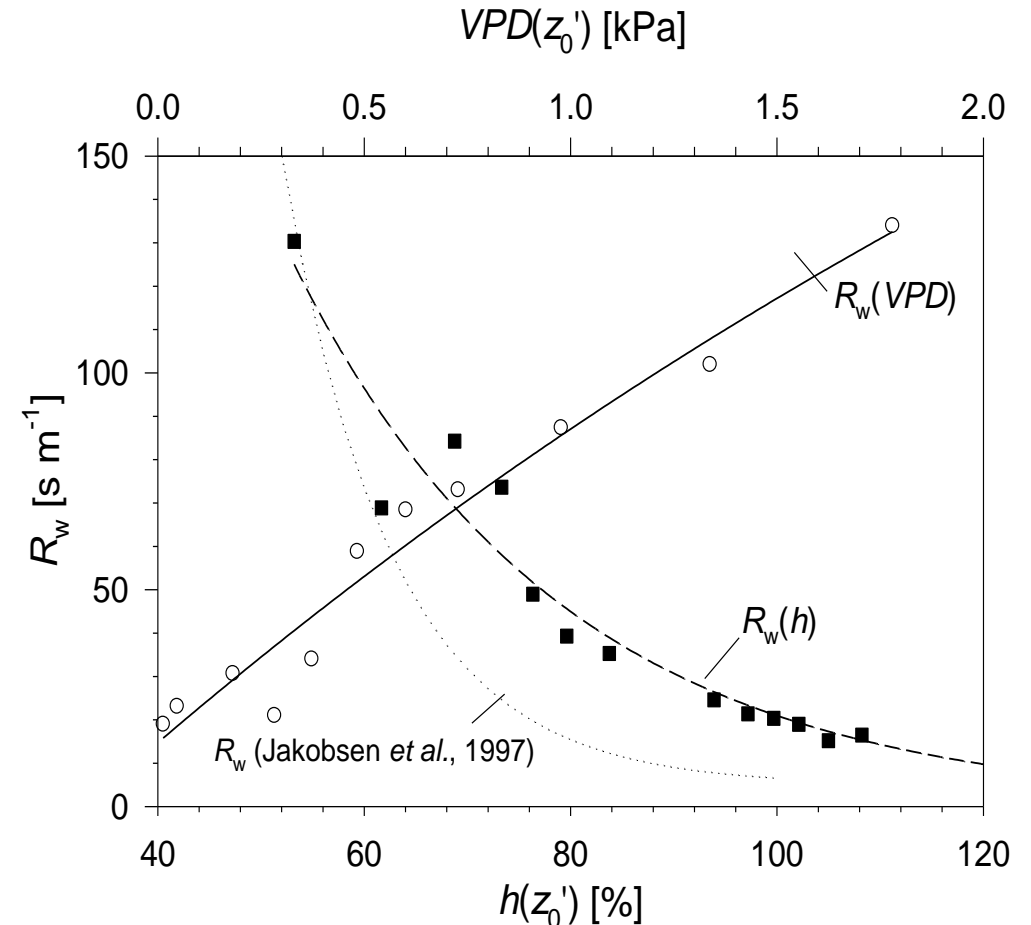
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# Determination of Cuticular Resistance ( $R_w$ )

- Derived as during dark (stomata closed) and windy conditions (e.g.  $R_a + R_b < 200 \text{ s m}^{-1}$ )

$$R_c = \chi_a / F_t - R_a - R_b$$

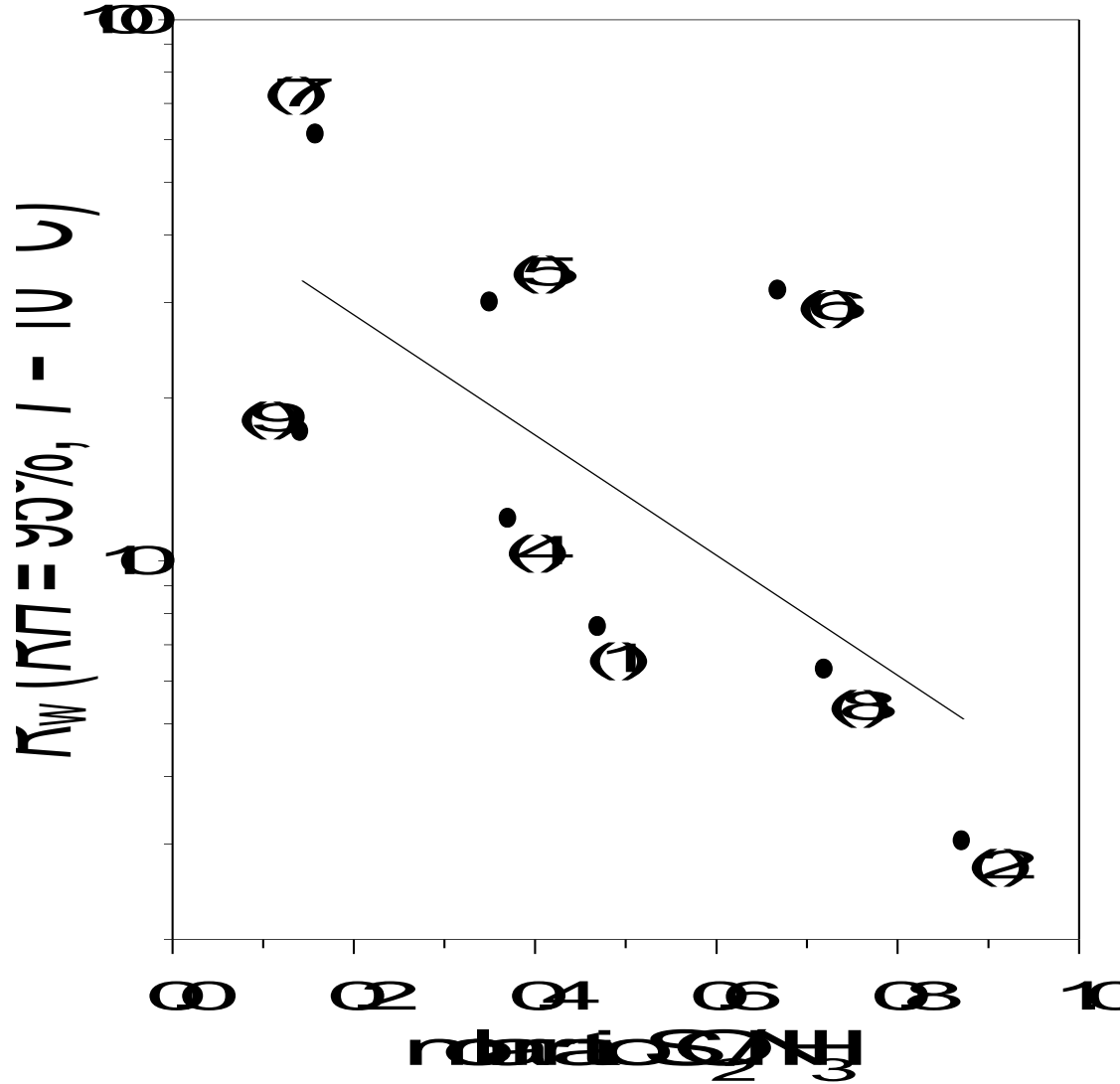
- $R_w$  is a function of chemical properties of cuticle and leaf water layers (Flechard *et al.* 1998).
- Vapour pressure deficit (VPD) or RH is often used to determine the parameterisation of  $R_w$  (Nemitz *et al.* 2000).



Taken from Nemitz *et al.* (2004) Surface exchange fluxes of  $\text{NH}_3$ ,  $\text{SO}_2$ ,  $\text{HNO}_3$  and  $\text{HCl}$  over a heathland. *Atmos. Chem. Phys.* 4, 489 - 1005.

The full squares and open circles represent block-averaged values for 50 sequential 30-min values of  $R_w$  values, sorted for  $h$  and VPD, respectively. The dashed and the solid lines show the parameterizations used to fit the measurements.

# Impact of the chemical climate on cuticular resistance ( $R_w$ )



Taken from Nemitz, Milford and Sutton (2000).  
 Quart. J. Royal Meteor:

Compilation of the cuticular resistance ( $R_w$ ) at relative humidity RH = 95%, and temperature, T = 10 °C, parametrized as a function of molar ratio of the mean SO<sub>2</sub> and NH<sub>3</sub> air concentrations. The numbers are those of the datasets used.

**Note:** a subsequent meta-analysis of Massa et al. (2010) considering a larger showed a more complex picture

SO<sub>2</sub> likely no longer the controlling acid gas. HNO<sub>3</sub> and HCl have risen in relative importance.

# Cuticular uptake resistance ( $R_w$ ) increases at high $\text{NH}_3$ concentrations (saturation effect)

## Ecosystem scale:

Jones et al., 2007; Concentration-dependent  $\text{NH}_3$  deposition processes for mixed moorland semi-natural vegetation Atmospheric Environment, 41 (2007), pp. 2049-2060

## For individual species response see:

Jones et al., 2007; Concentration-dependent  $\text{NH}_3$  deposition processes for moorland plant species with and without stomata. Atmospheric Environment, 41, pp 8980-8994

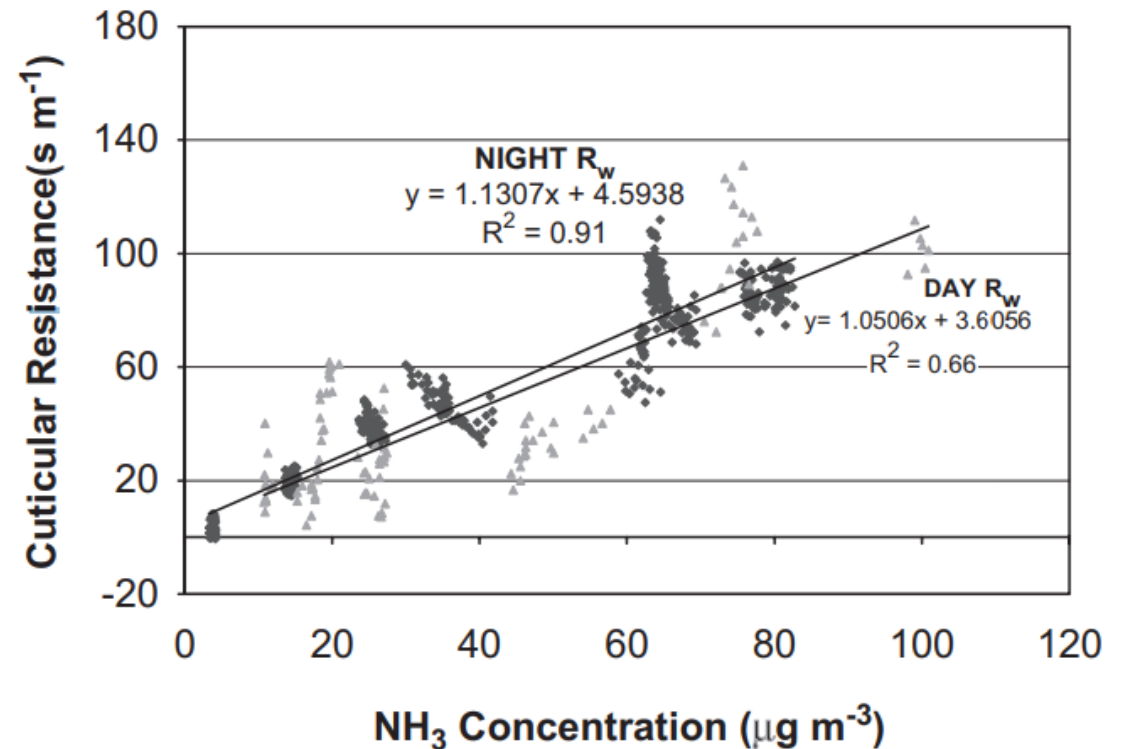
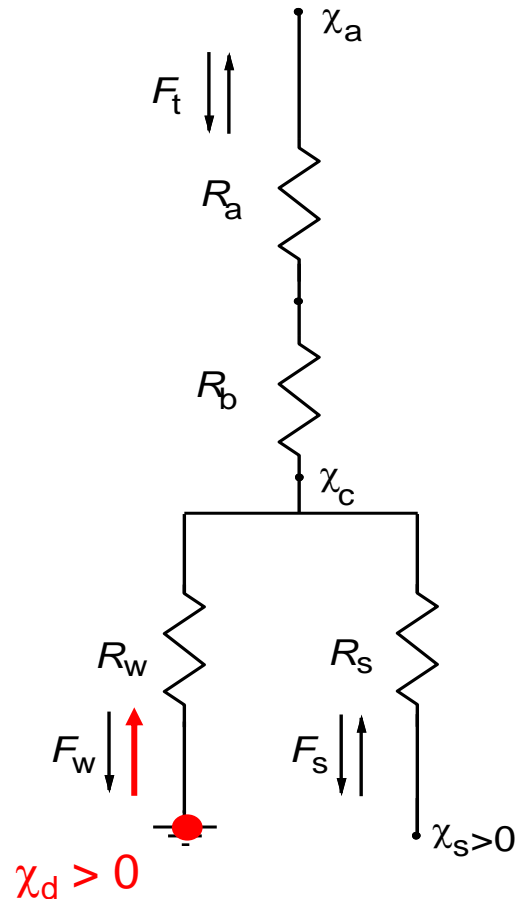


Fig. 10. Relationship between  $\text{NH}_3$  concentrations ( $\mu\text{g m}^{-2} \text{s}^{-1}$ ) and daytime  $R_w$  ( $\text{s m}^{-1}$ ) (light grey triangles) and night-time  $R_w$  ( $\text{s m}^{-1}$ ) (black diamonds) for a mixed vegetation canopy.



# Leaf surfaces respond to chemical composition and can saturate at high $\text{NH}_3$ levels



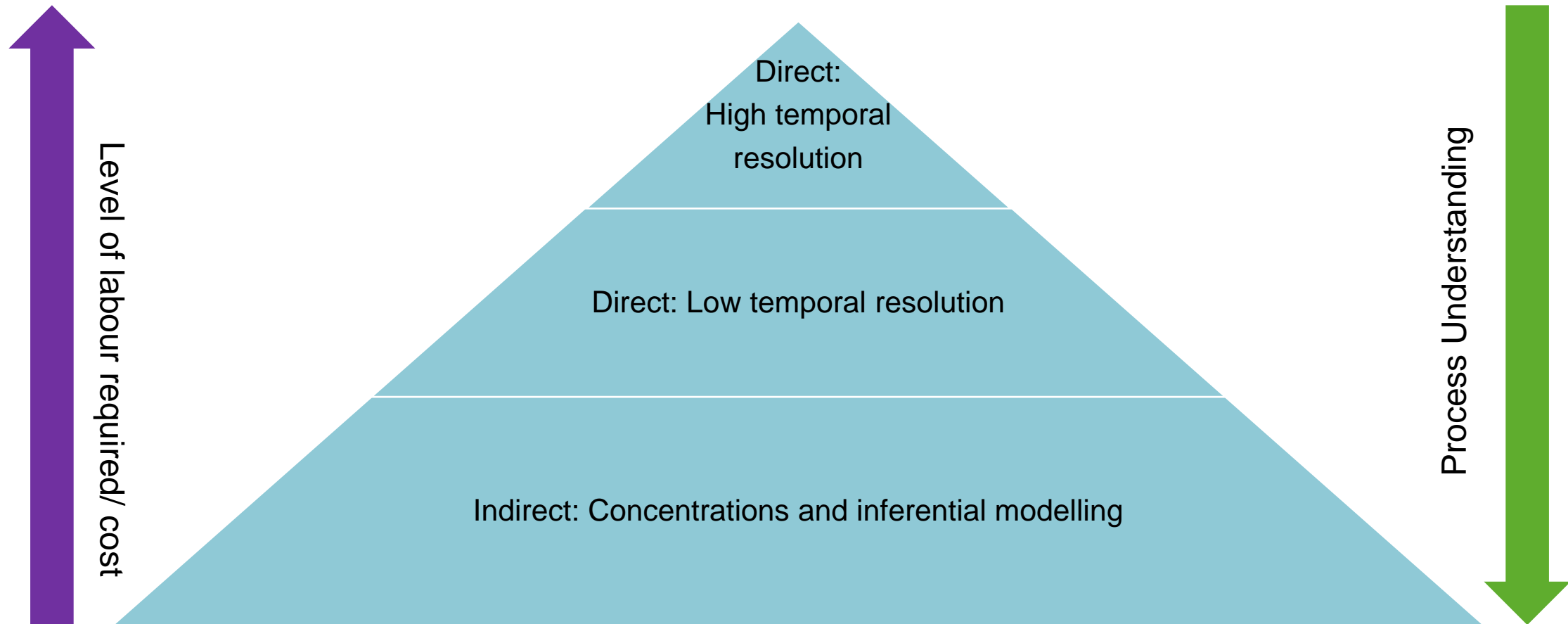
- Increase in  $R_w$  or introduction of additional cuticular emission potential ( $\chi_d$ )

Two examples of approaches to deal with  $R_w$  in ACTMs:

- EMEP:  $R_w$  function of annual average  $\text{NH}_3/\text{SO}_2$  concentration ratio in air
- DEPAC (LOTOS-EUROS):  $R_w$  as a function of ambient  $\text{NH}_3$  concentration (Jones et al., 2007a, b)

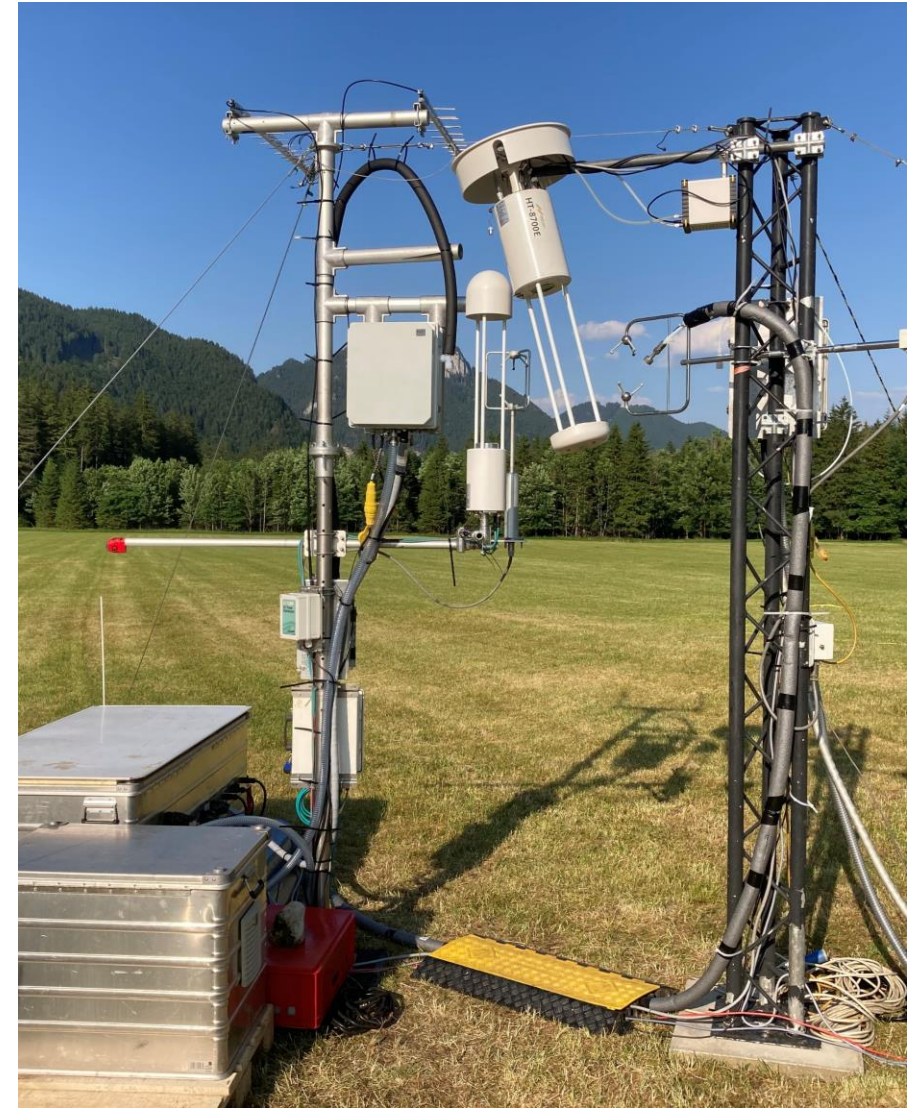
➤ In both models only deposition to the cuticle is possible

# NH<sub>3</sub> dry deposition measurement approaches



# Direct measurements

- Micrometeorological methods have to fulfil requirements
  - homogenous flat terrain
  - sufficient turbulence
  - Eddy covariance ‘ideal method’ requires sampling of up  $\geq 10$  Hz
- Challenging under ambient conditions due to ‘sticky’ nature of  $\text{NH}_3$
- Recently there now open path systems on the market to undertake direct flux measurements
- Not applicable near point sources (due to advection errors)

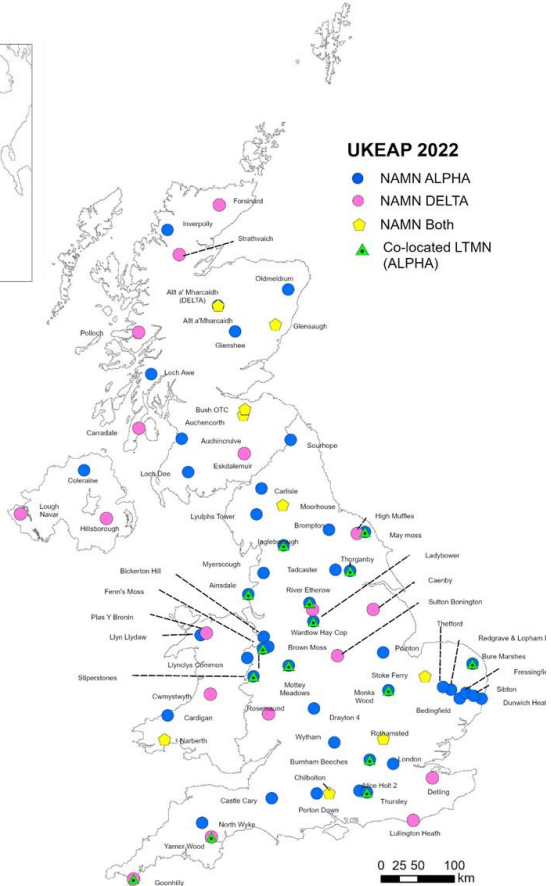
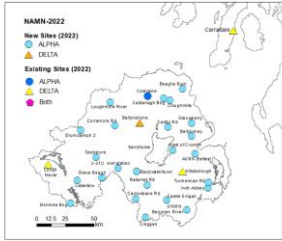


# Indirect measurements

- Two types of concentration measurements:
  - State-of-the art high temporal resolution (20 Hz to 1 hour) at 1 location
  - Multiple sensors with low temporal resolution covering a spatial area (daily to monthly average)
- Data from the measurements are then used to inform models:
  - Measurement model fusion approach (Concentration Based Estimate Deposition Model, CBED<sup>1</sup>)
  - Calibration of an atmospheric chemistry transport model (UK: EME4UK)<sup>2</sup>
  - Use similar parameterisations as previously shown



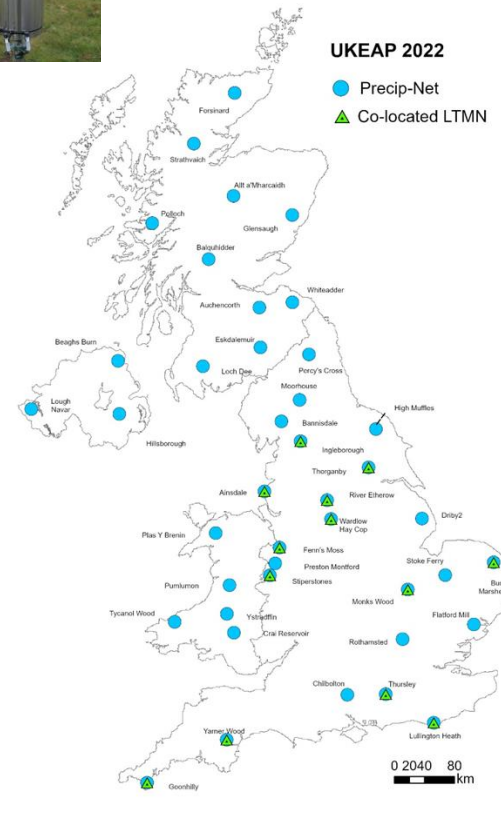
# UK Monitoring Networks



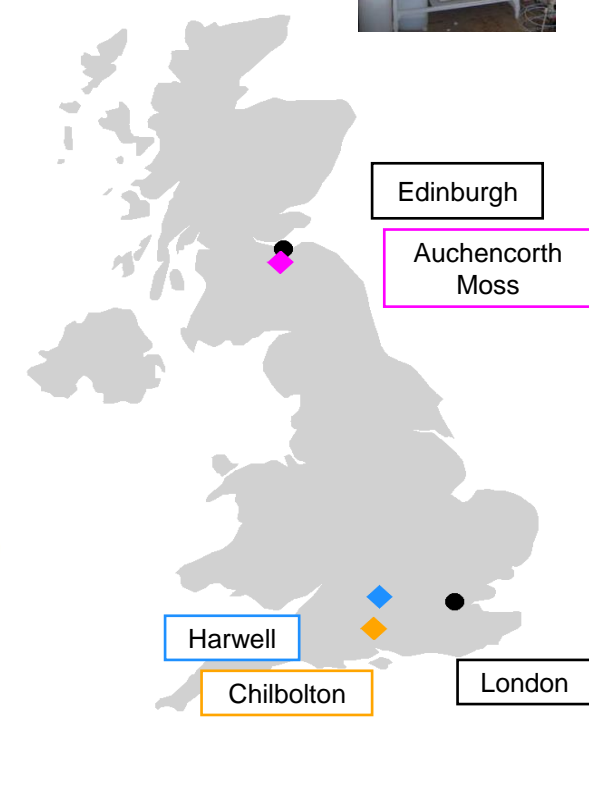
National Ammonia Monitoring Network (112 sites)



Acid gases and Aerosols Network (28 sites)



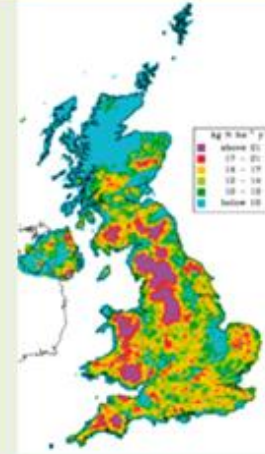
Precipitation Network (41 sites)



EMEP supersites (2 sites)

# UK Monitoring Networks

Modelling and mapping pollutant concentrations and deposition



National and international assessments atmospheric pollution and deposition to the environment

Critical Loads and exceedance mapping

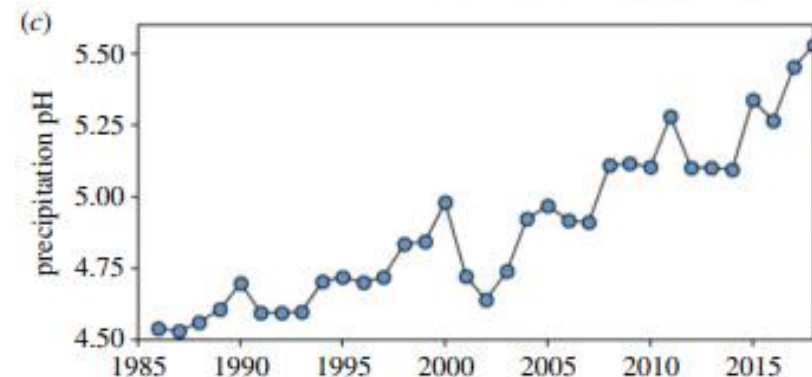
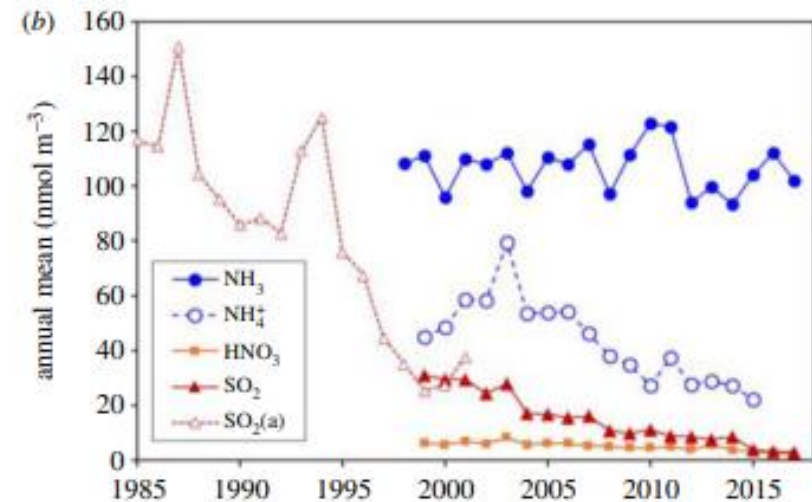
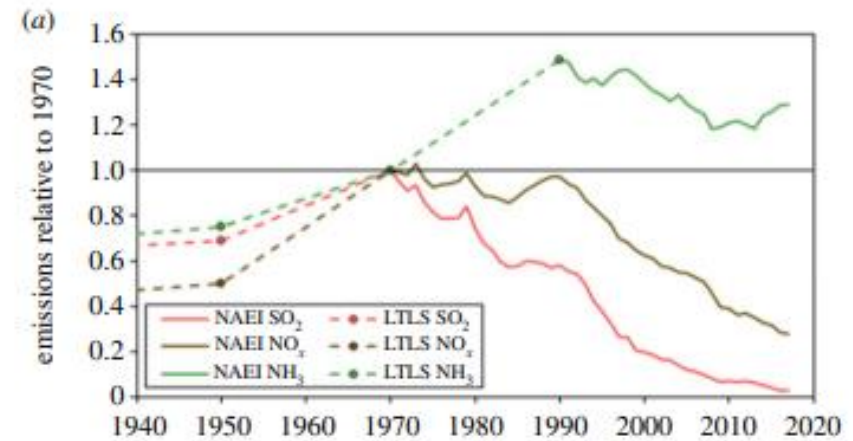


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# Current status of NH<sub>3</sub> in the UK

- Decrease in NH<sub>3</sub> emissions is not reflected in NH<sub>3</sub> concentrations
- UK moving from an acidic towards an alkaline environment
- Likelihood that R<sub>w</sub> is increasing and more potential for desorption from the cuticle



Taken from Sutton, M.A. *et al.* 2020. Alkaline air: changing perspectives on nitrogen and air pollution in an ammonia-rich world. *Philosophical Transactions of the Royal Society A*, 378(2183), p.20190315.

# Impact of NH<sub>3</sub> currently in the UK



- Critical levels (annual)
  - >1 & ≤ 3 μg m<sup>-3</sup> (lichens and byrophytes exceeded)
  - > 3 μg m<sup>-3</sup> (vascular plants)
- 69% of the UK is currently exceeding the critical levels for NH<sub>3</sub> set for lichens and byrophytes
- Areas of high concentrations of NH<sub>3</sub> often associated with local intensive agricultural areas

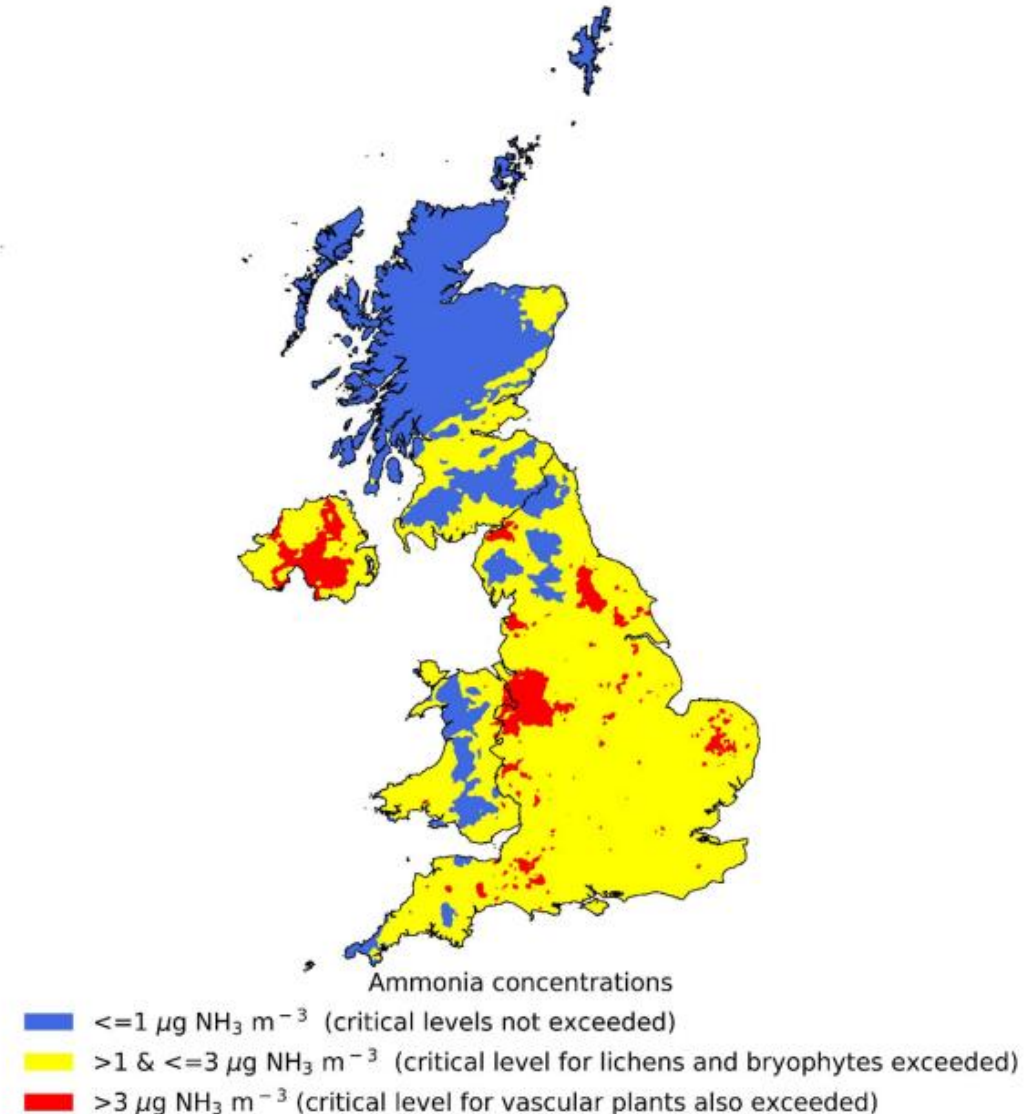


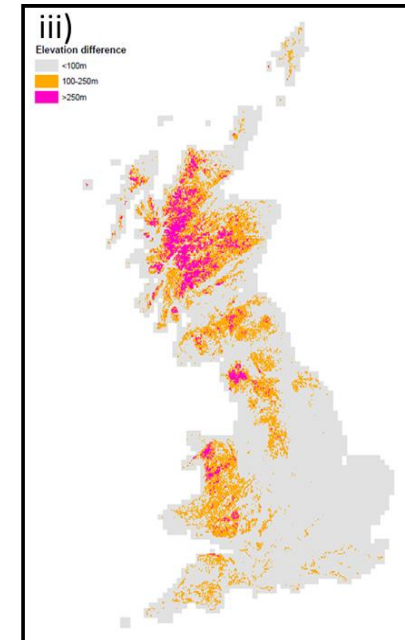
Figure 4.1: EMEP 1 x 1 km mean ammonia concentrations for 2017-19.



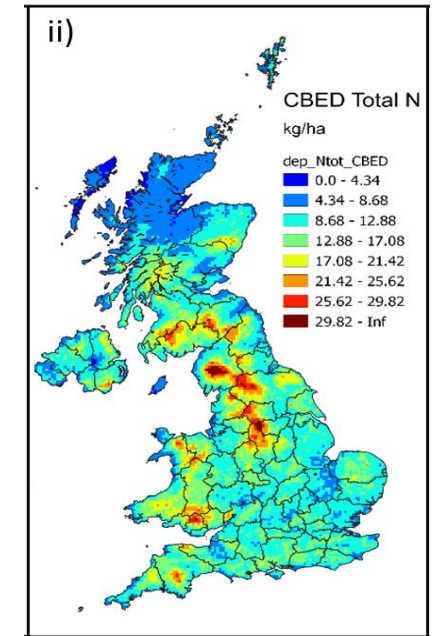
# Challenges: quantifying dry deposition to complex terrain

- Dry deposition parameterisations derived for homogeneous, flat surface
- Micrometeorological methods not applicable
- Models currently do not agree the deposition rates to complex terrain
- Expected enhancement of deposition rates to complex terrains
  - Vulnerable ecosystems to N deposition are at risk

Definition of complex terrain

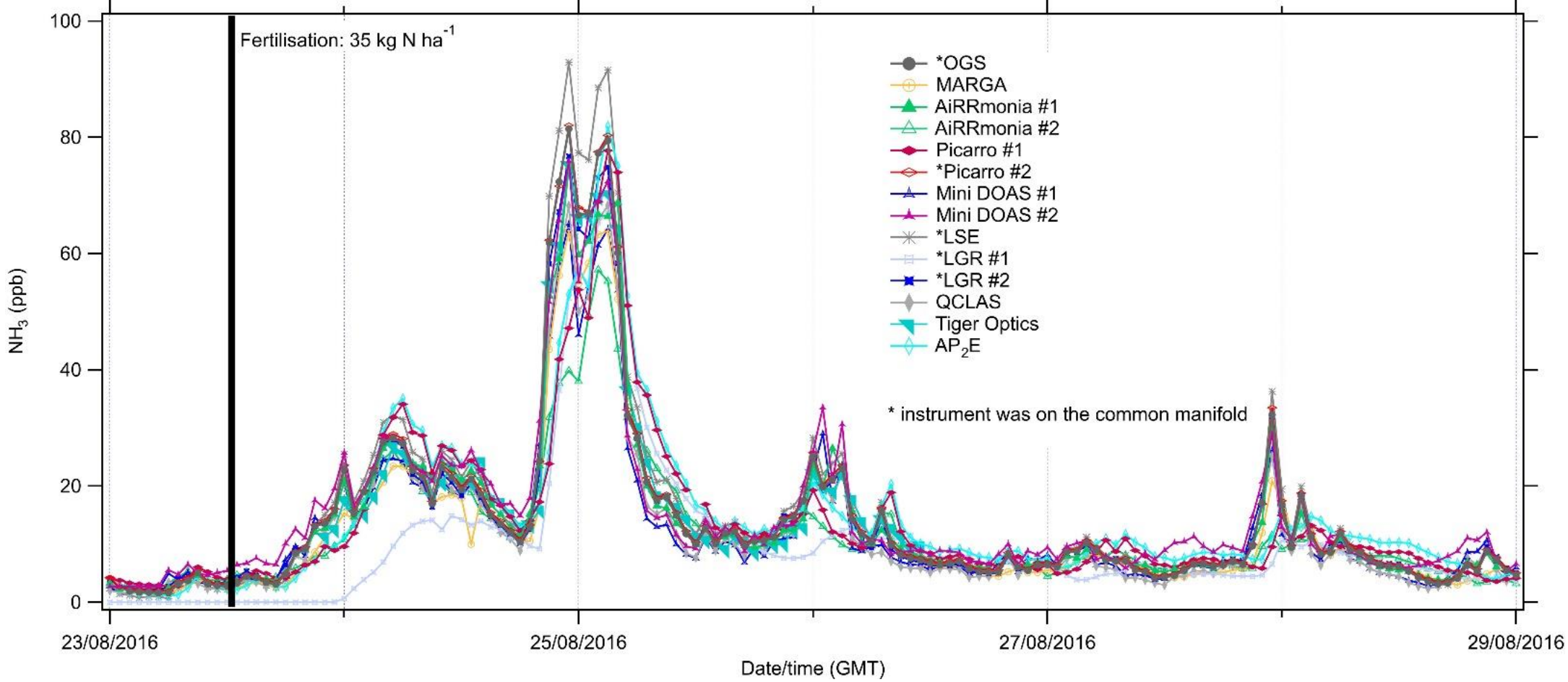


Coincides with large  $N_{dep}$



**Taken from:** Cowan, N., Nemitz, E., Walker, J.T., Fowler, D., Finnigan, J.J., Webster, H.N., Levy, P., Twigg, M., Tang, S.Y., Bachiller-Jareno, N. and Trembath, P., 2022. Review of methods for assessing deposition of reactive nitrogen pollutants across complex terrain with focus on the UK. *Environmental Science: Atmospheres*.

# Challenges: Comparability of NH<sub>3</sub> concentration measurements



# Ammonia as a future fuel: some considerations

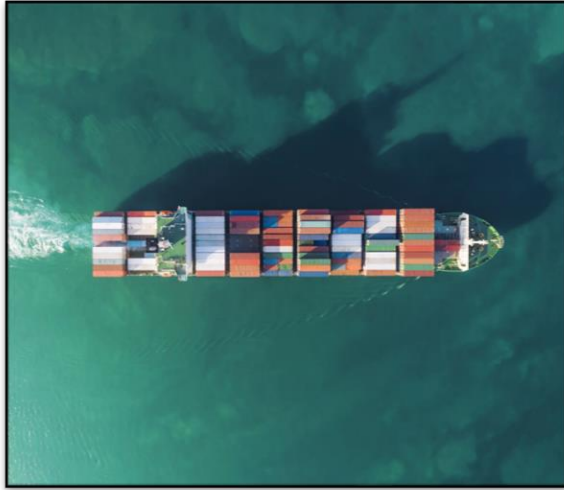


Photo credit: [www.faam.ac.uk](http://www.faam.ac.uk)

- High concentrations are likely to significantly increase the cuticular resistance or introduce a cuticular emission potential ( $\chi_d$ )
  - Leads to an increase in atmospheric lifetime due to reduction in deposition velocity
  - Further studies are required to determine if an emission potential occurs on the cuticle at elevated concentrations across different environments



# Ammonia as a future fuel: some considerations

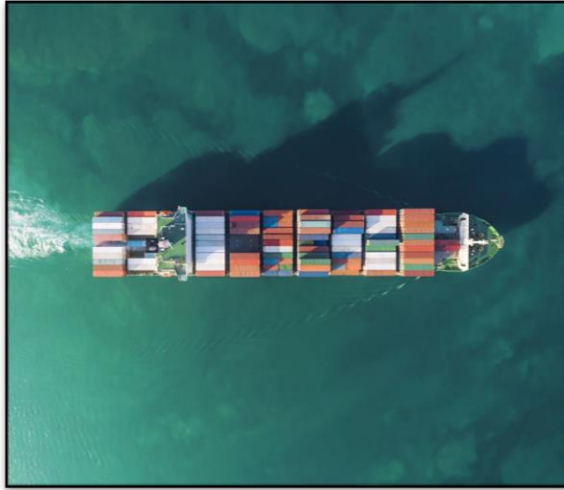


Photo credit: [www.faam.ac.uk](http://www.faam.ac.uk)

- Incident capability in the UK:
  - National monitoring network can help in identify any changes in the state of the UK background concentrations but won't give process understanding
  - A system needs to be developed to quantify dispersion of  $\text{NH}_3$  in response to any incident in the future



# Thank you

For more information  
please contact:

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UK Centre for  
Ecology & Hydrology

