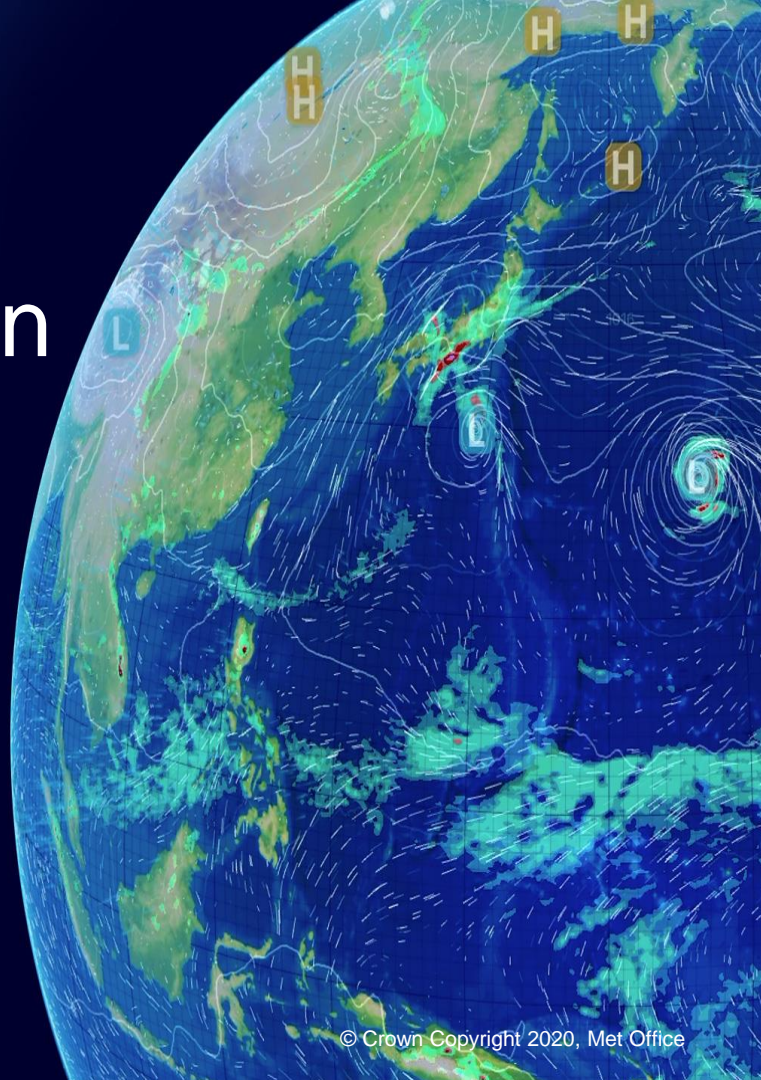
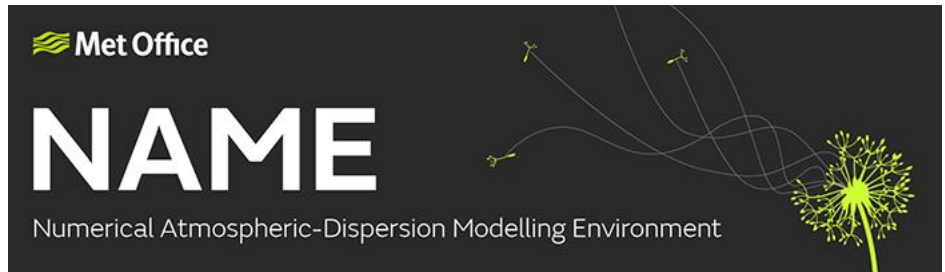


Modelling dry deposition in an operational Lagrangian model

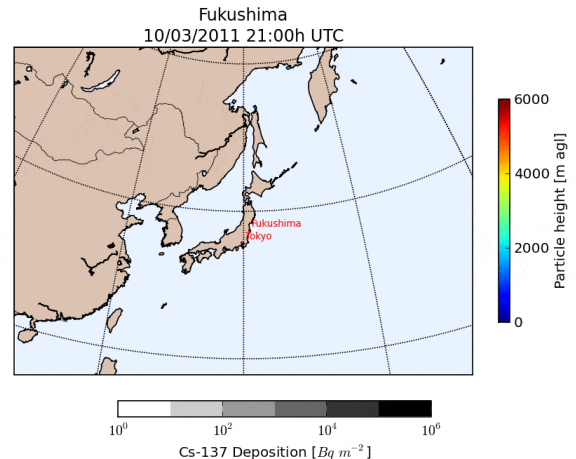
Helen Webster





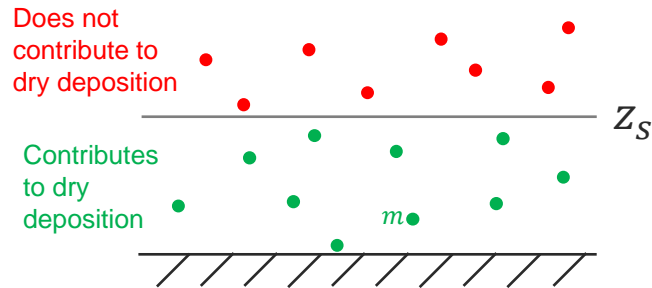
Predicts the atmospheric transport **and deposition to the ground surface** of airborne substances

- Offline Lagrangian model
 - Eulerian sub-model
- Used operationally
- Developed following 1986 Chernobyl accident
 - Now has a wide range of atmospheric dispersion applications
 - Gases & particles... and even insects



Dry deposition parameterisation

- Uses the concept of a deposition velocity v_d
- Flux of material to ground is proportional to concentration
 - $F = v_d C$
- Dry deposition
 - $\Delta m = m \left[1 - \exp \left(-\frac{v_d}{z_s} f \Delta t \right) \right]$
 - $0 \leq f \leq 1$, fraction of timestep $z \leq z_s$
- All Lagrangian particles below the deposition height z_s dry deposit
 - Default z_s - boundary layer height



Deposition velocities

Increasing complexity 

User specified

- Simple
- v_d fixed
- Specie dependent
- Based on literature reported deposition values

Resistance analogy

- $$v_d = \frac{1}{R_a + R_b + R_c}$$
- Meteorological dependency
- Fixed specie-dependent surface resistance

Land use dependent scheme

- Resistance analogy
- Varying surface resistance dependent on land-surface
- Selected species

Resistance analogy

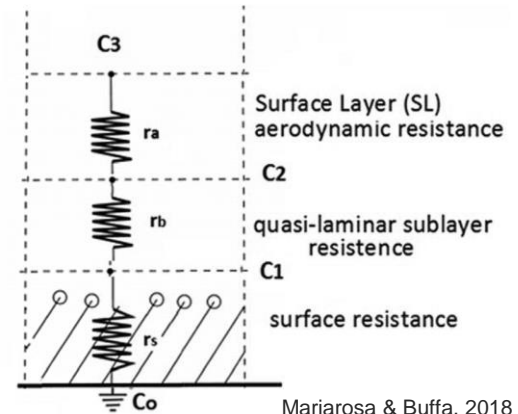
- Aerodynamic resistance depends on stability and roughness length

$$R_a = \frac{1}{ku_*} \left[\ln \left(\frac{z_r + z_0}{z_0} \right) - \Phi \right],$$

where

$$\Phi = \begin{cases} 2 \ln \left(\frac{1 + \left(1 - 16 \left(\frac{z_r + z_0}{L} \right)^{1/2} \right)}{1 + \left(1 - 16 \left(\frac{z_0}{L} \right)^{1/2} \right)} \right), & \text{unstable,} \\ -5 \frac{z_r}{L}, & \text{stable } \frac{z_r + z_0}{L} < 1, \\ -\frac{5}{L} (L - z_0) - 5 \ln \left(\frac{z_r + z_0}{L} \right), & \text{stable } \frac{z_0}{L} < 1 \leq \frac{z_r + z_0}{L}, \\ -5 \ln \left(\frac{z_r + z_0}{z_0} \right), & \text{stable } \frac{z_0}{L} \geq 1. \end{cases}$$

Meteorological dependency



Mariarosa & Buffa, 2018

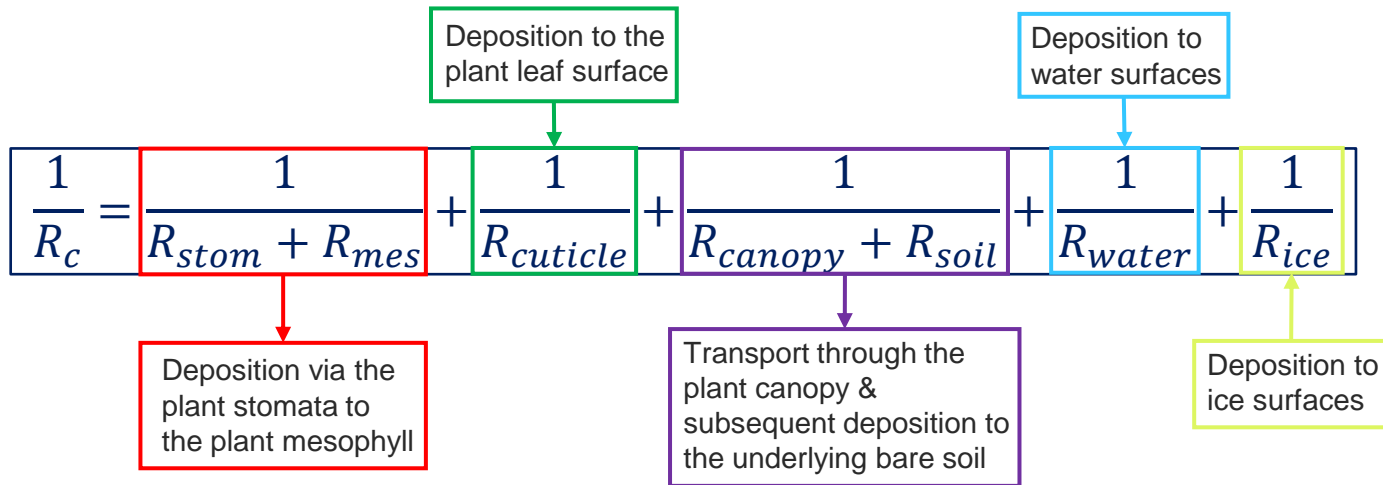
$$v_d = \frac{1}{R_a + R_b + R_c}$$

- Laminar resistance depends on friction velocity

$$\bullet R_b = \frac{1}{8u_*}$$

Land-use dependent scheme

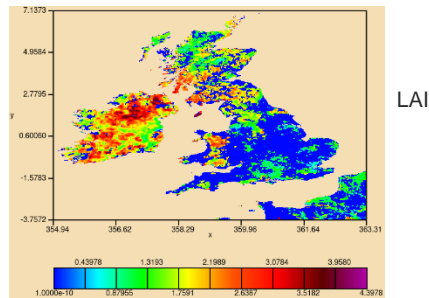
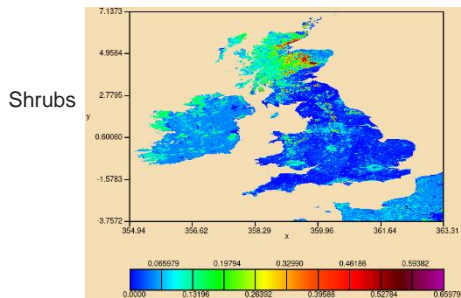
- Uptake at the ground is strongly dependent on the land-surface type & properties



STOCHEM parameterisations

ozone, nitric oxide, nitrogen dioxide, nitric acid, PAN, hydrogen peroxide, methane, carbon monoxide, hydrogen, formaldehyde, sulphur dioxide, ammonia

- Requires additional surface and plant information
 - e.g., stomatal conductance, leaf area index
 - Ancillaries: fixed (e.g., land use) or monthly (e.g., LAI)
 - NWP data: varying with time (e.g., stomatal conductance)

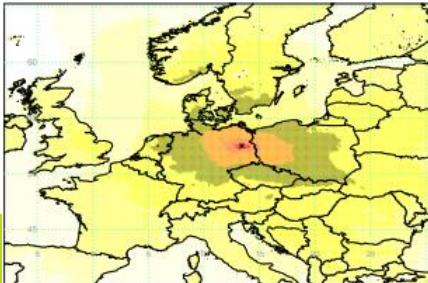


STOCHEM parameterisations

ozone, nitric oxide, **nitrogen dioxide**, nitric acid, PAN, hydrogen peroxide, methane, carbon monoxide, hydrogen, formaldehyde, sulphur dioxide, ammonia

- Deposition to vegetation occurs mainly via the stomata
- Mesophyllic resistance is **not** negligible

land-sea effect:
less deposition
over water than
over land



$$\frac{1}{R_c} = \frac{1}{R_{stom} + R_{mes}} + \frac{1}{R_s} = \frac{g_c}{2.4} + \frac{1}{R_s}$$

g_c - Stomatal conductance

Land-surface index	Land-surface type	Standard surface resistance (s m ⁻¹)	R_s
1	Broadleaf trees	225	
2	Needleleaf trees	225	
3	C3 (temperate) grass	400	
4	C4 (tropical) grass	400	
5	Shrubs	600 (steppe), 1200 (tundra)	
6	Urban	1200	
7	Inland water	2600	
8	Bare soil	1200 (steppe and tundra)	
9	Land-ice	3500	

Table 3: Standard surface resistances for nitrogen dioxide

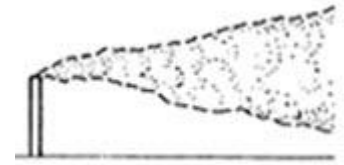
STOCHEM parameterisations

ozone, nitric oxide, nitrogen dioxide, nitric acid, PAN, hydrogen peroxide, methane, carbon monoxide, hydrogen, formaldehyde, sulphur dioxide, ammonia

- Deposits to vegetation via the stomata and the cuticle
- The dry deposition rate depends on the wetness of the leaf surface
- The cuticular resistance parameterisation has a dependence on surface temperature (T_s) and canopy water (cw)

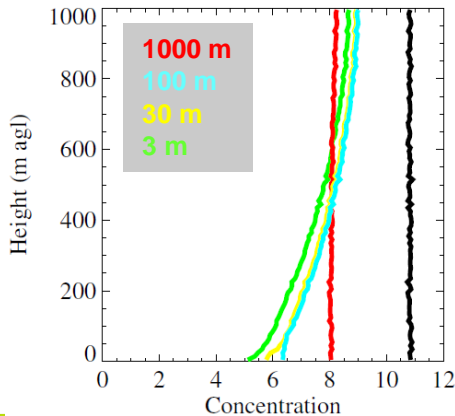
$$\frac{1}{R_c} = \frac{1}{R_{stom}} + \frac{1}{R_{cuticle}} + \frac{1}{R_s} = \frac{g_c}{0.97} + \frac{1}{R_{cuticle}} + \frac{1}{10}$$

$$R_{cuticle} = \begin{cases} 1000.0, & T_s < 268 \text{ K} \\ 200.0, & 268 \text{ K} \leq T_s \leq 272 \text{ K} \\ 10.0, & T_s > 272 \text{ K} \ \& \ cw > 0.25 \text{ mm} \\ 5.0 \log [T_s - 271.15] \exp \left[\frac{100.0 - rh}{12.0} \right], & T_s > 272 \text{ K} \ \& \ cw \leq 0.25 \text{ mm} \end{cases}$$



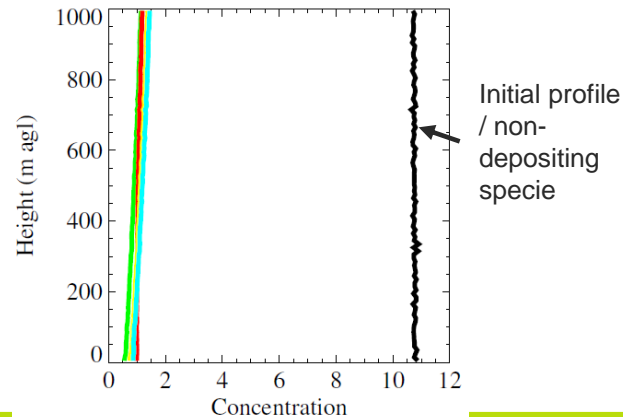
Near-source modelling

- Material is not well mixed within the boundary layer
- User option to specify a lower deposition height z_s
 - Noisy deposition fields (fewer Lagrangian particles)
 - Increased run-time (more Lagrangian particles required)



(a) 6 hours

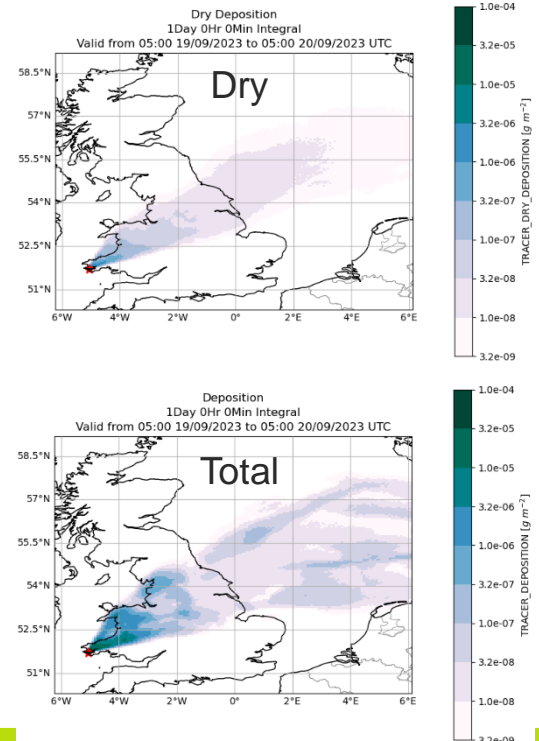
Well mixed



(b) 36 hours

Model validation of dry deposition

- Difficult
- Lack of deposition observations
 - Often long-term measurements
 - Total deposition (including wet)
 - For soluble species, wet deposition often dominates
- Often validated indirectly via air concentrations
 - Uncertainties in emissions, transport, dispersion, chemistry, wet deposition, etc.
 - Tracer experiments: Kincaid, ETEX, CAPTEX / ANATEX
 - Events: air quality, fires, volcanic eruptions, radiological incidents, etc.



Validation

Atmospheric Environment 119 (2015) 131–143

Contents lists available at ScienceDirect

Atmospheric Environment

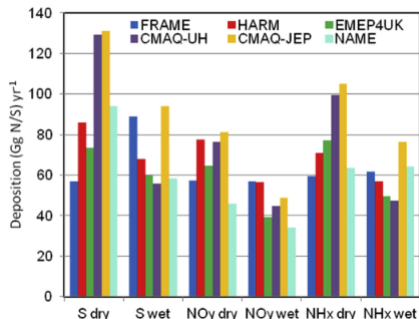
journal homepage: www.elsevier.com/locate/atmosenv



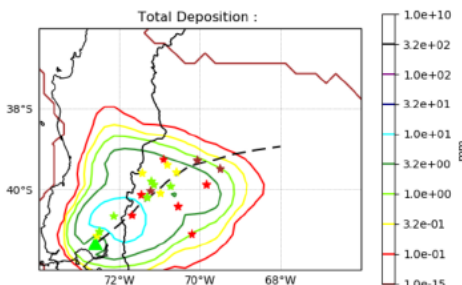
Evaluation of the performance of different atmospheric chemical transport models and inter-comparison of nitrogen and sulphur deposition estimates for the UK



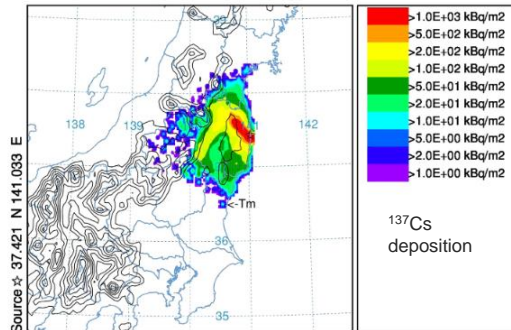
A.J. Dore^{a,*}, D.C. Carslaw^b, C. Braban^a, M. Cain^{c,d}, C. Chemel^e, C. Conolly^f, R.G. Derwent^g, S.J. Griffiths^h, J. Hallⁱ, G. Hayman^j, S. Lawrence^c, S.E. Metcalfe^k, A. Redington^l, D. Simpson^{m,n}, M.A. Sutton^a, P. Sutton^o, Y.S. Tang^a, M. Vieno^a, M. Werner^p, J.D. Whyatt^q



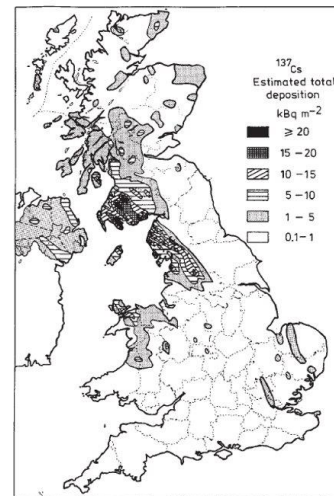
2015 Calbuco eruption



Fukushima



Chernobyl



Clark & Smith, 1988

Future work - UKCA

Geosci. Model Dev., 13, 1223–1266, 2020
<https://doi.org/10.5194/gmd-13-1223-2020>
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Geoscientific
Model Development 

Description and evaluation of the UKCA stratosphere–troposphere chemistry scheme (StratTrop v1.0) implemented in UKESM1

Alexander T. Archibald^{1,2}, Fiona M. O'Connor³, Nathan Luke Abraham^{1,2}, Scott Archer-Nicholls¹, Martyn P. Chipperfield^{1,5}, Mohit Dalvi³, Gerd A. Folberth³, Fraser Dennison^{6,a}, Sandip S. Dhomse^{4,5}, Paul T. Griffiths^{1,2}, Catherine Hardacre³, Alan J. Hewitt³, Richard S. Hill³, Colin E. Johnson³, James Keeble^{1,2}, Marcus O. Köhler^{1,7,b}, Olaf Morgenstern⁶, Jane P. Mulcahy³, Carlos Ordóñez^{3,c}, Richard J. Pope^{4,5}, Steven T. Rumbold⁸, Maria R. Russo^{1,2}, Nicholas H. Savage³, Alistair Sellar³, Marc Stringer⁸, Steven T. Turnock³, Oliver Wild⁹, and Guang Zeng⁶

- Implementation of UKCA chemistry scheme into NAME
 - Includes dry deposition scheme
 - Uses resistance analogy
 - Similar to land-use dependent scheme
 - Widely used
 - Frequent updates (e.g., recently SO₂, O₃)

Questions?